

PHYSICS 114 Contemporary Physics II – Homework 2

In all problems involving solutions to problems involving equations, leave your solution in “equation” form until the last step.

1. Suppose we have two blocks composed of 5 harmonic oscillators each and have 4 quanta in each initially. After bringing them together, what is the probability of having all the energy move to one of the blocks?

A useful approximation for calculating entropy of a macrostate is the stirring approximation:

$$\ln N! = N \ln N - N \quad (1)$$

2. Suppose we have a block consisting of N oscillator
 - (a) Write down the equation for the number of microstates as a function of the number of quanta q that we put into the block.
 - (b) Write down an expression for the entropy using the stirring approximation .
 - (c) Write down an expression for the entropy in terms of energy E . (That is use the expression $E = q\hbar\omega$).
 - (d) Write down an analytical expression (by taking the derivative of the above expression) for the temperature.
3. Probable energy states for molecules at room temperature
 - (a) Suppose we have an system that has the following allowed electronic energy states: 1 eV, 1.5 eV, 3.0 eV, 10eV. What is the probability of the system to be at each of these energy states at 298K, 1000K?
 - (b) Suppose we have a system that has the following allowed rotational energy states: $1.1 * 10^{-3}$ eV, $1.3 * 10^{-3}$, $1.5 * 10^{-3}$ eV. What is the probability of populating these rotational states at 300K and 5000K?
 - (c) From the above analysis what rotational and electronic states are you likely to find a typical molecule in at room temperature?

Problems 11: P.54 (do this first), P.43, P.51, P.52